St. Andrews Scots Sr. Sec. School

9th Avenue, I.P. Extension, Patparganj, Delhi – 110092 Session: 2024 – 2025 (Worksheet-4)

Class:	VIII	Subject: Science	Chapter: Combustion and flame	
Q.1. D	efine:			
,		able substances		
,	b) Ignition temperature			
,	c) Soot			
d) Calorific value				
Q.2.Fi	II in the I	blanks:		
a)	The burning of substance takes place when oxygen is present then it is called			
b)	Substances that don't catch fire in the presence of oxygen are called			
c)	The minimum temperature required by any substance to catch fire or burn is called			
d)	The flame of Bunsen burner becomes blue due to the sufficient supply of			
e)	And substances are examples of the explosion.			
Q.3. N	lultiple C	Choice Questions:		
	Which one of the following has the highest calorific value?			
	(a) Kerosene (b) Biogas (c) LPG			
b)	If a person's clothes catch fire, the best way to extinguish the fire is to			
	(a) throw water on the clothes			
	(b) use fire extinguisher(c) cover the person with a woollen blanket.			
c)	The calorific value of a fuel is expressed in a unit called			
	(a) kilojoules per litre			
	(b) Kilogram per millilitre			
		ules per gram		
d)		es, people use wood as f	uel because	
•	(a) it is considered to be an ideal fuel			

Q.4. State whether the following statement is true or false.

(b) Of its easy availability and low cost

(c) It is environment friendly

- a) A physical process in which a substance reacts with oxygen to give off heat is called combustion
- b) Increased concentration of nitrogen in air is believed to cause global warming.

- c) Middle zone is the hottest zone of a flame.
- d) The substances which vaporize during burning, give the flame

Q.5. Short Questions Answers:

- a) Why sodium is kept immersed in kerosene?
- b) Why isn't hydrogen gas used as a domestic or industrial fuel, although it has a very high calorific value?
- c) Although wood has a very high calorific value, we still discourage its use as a fuel. Explain.
- d) Can the process of cellular respiration be called combustion? Why?